

Wetting, Spreading & Adhesion: Lecture 2

Capillarity: Measurements

by Rossen Sedev

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Ian Wark Research Institute

Australian Research Council Special Research Centre
For Particle and Material Interfaces

The *Wark*™

Outline

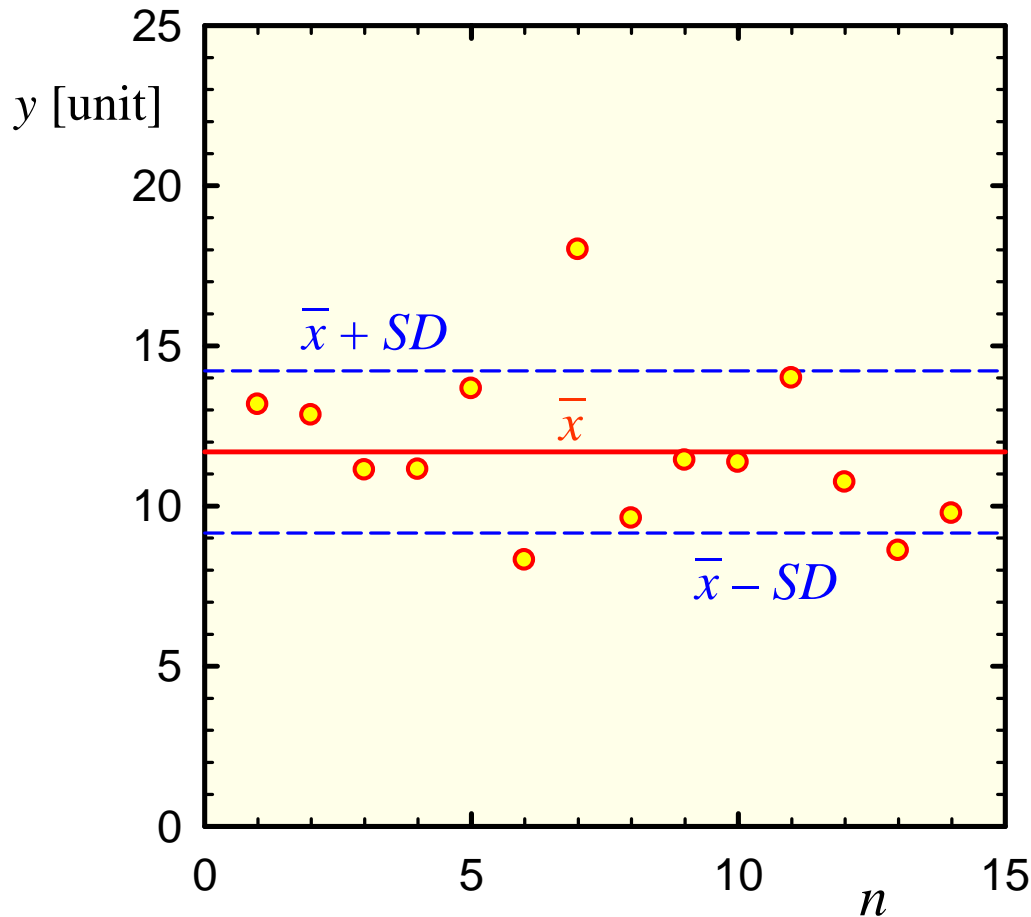
- **Physical Measurements**
 - Errors
 - Accuracy & Precision
 - Presentation
- **Surface Tension Measurement**
 - Pendant Drop (shape measurement)
 - Wilhelmy Plate (force measurement)
 - Maximum Bubble Pressure
- **Contact Angle Measurement on Flat Surfaces**
 - Sessile Drop (shape measurement)
 - Wilhelmy Plate (force measurement)
- **Contact Angle Measurement on Powders**
 - Skin Flotation
 - Static Capillary pressure
 - Washburn Method

1. Physical Measurements

- Mean & Standard Deviation
- SD versus SE
- Error Propagation
- Precision & Accuracy

Mean & Standard Deviation

Consider n identical and independent measurements (x_i , $i = 1, \dots, n$):



$$\bar{x} = \frac{1}{n} \sum_{i=1}^n x_i$$
$$SD = \sqrt{\frac{1}{n-1} \sum_{i=1}^n (x_i - \bar{x})^2}$$

Standard Deviation or Standard Error?

There are different measures of the data spread:

Standard Deviation: $x = m \pm SD$

Gives a 68% probability of finding a single measurement

Standard Error: $x = m \pm \frac{SD}{\sqrt{n}}$

Gives a 68% probability of finding an average measurement

Confidence Interval: $x = m \pm t_{\alpha/2, n-1} \frac{SD}{\sqrt{n}}$

Gives a $(1-\alpha)$ probability of finding an average measurement

Any option is valid as long as it is properly quoted!

How Good Is a Measurement?

Best Value

Absolute Error

Relative Error

$$y = \bar{y} \pm \Delta y$$

$$\varepsilon = \frac{\Delta y}{y}$$

Value	Error	Relative Error [%]	Quality
10.0	0.5	5	OK (maybe)
500.0	0.5	0.1	Good
0.5	0.5	100	Unacceptable

Error Propagation

The values, a and b , and their errors, Δa and Δb , are known. What is the error of y ?

Relation	Maximum Error	Most Probable Error
$y = a + b$	$\Delta y = \Delta a + \Delta b$	$(\Delta y)^2 = (\Delta a)^2 + (\Delta b)^2$
$y = a - b$	$\Delta y = \Delta a + \Delta b$	$(\Delta y)^2 = (\Delta a)^2 + (\Delta b)^2$
$y = ab$	$\Delta y/y = \Delta a/a + \Delta b/b$	$(\Delta y/y)^2 = (\Delta a/a)^2 + (\Delta b/b)^2$
$y = a/b$	$\Delta y/y = \Delta a/a + \Delta b/b$	$(\Delta y/y)^2 = (\Delta a/a)^2 + (\Delta b/b)^2$

The general case:

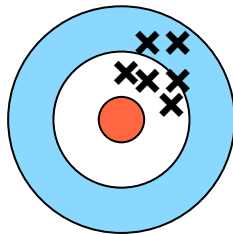
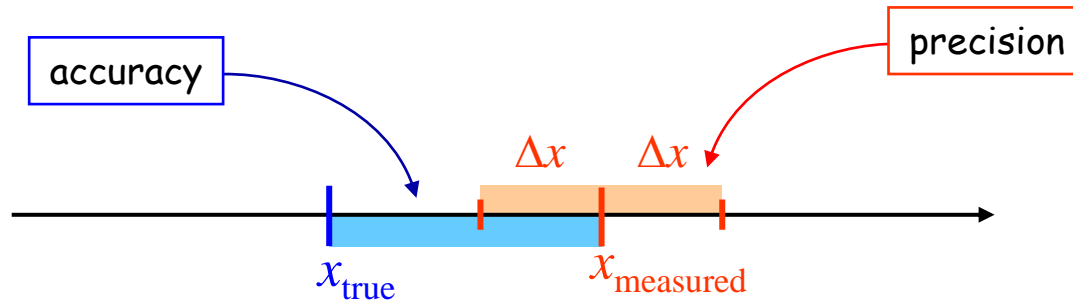
$$y = f(a, b)$$

$$dy = \frac{\partial f}{\partial a} da + \frac{\partial f}{\partial b} db$$

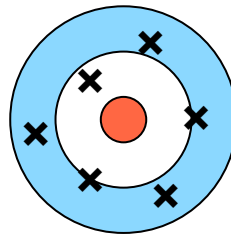
$$\Delta y = \left| \frac{\partial f}{\partial a} \right| \Delta a + \left| \frac{\partial f}{\partial b} \right| \Delta b$$

$$\Delta y^2 = \left(\frac{\partial f}{\partial a} \right)^2 \Delta a^2 + \left(\frac{\partial f}{\partial b} \right)^2 \Delta b^2$$

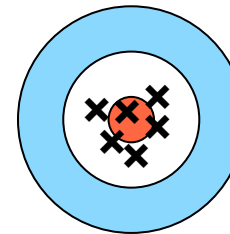
Precision & Accuracy



precise



accurate



precise & accurate

	Origin	Problem	Solution
Precision	random errors	lack of reproducibility	repeat measurements
Accuracy	systematic errors	lack of standard	use complementary techniques

Data Presentation

284.7 ± 3.2936

x

- How many measurements (e.g. 3 or 300)?
- Standard error, standard deviation or confidence interval?
- Rounding off the error:

3.293617 → 3

29.01 → 30

8.234 → 8

1.34 → 1.3

- Rounding off the average:

✓

285 ± 3

285 (3)

25.2 ± 0.3

25.2 (3)

6.83 ± 0.07

6.83 (7)

2. Surface Tension Measurement

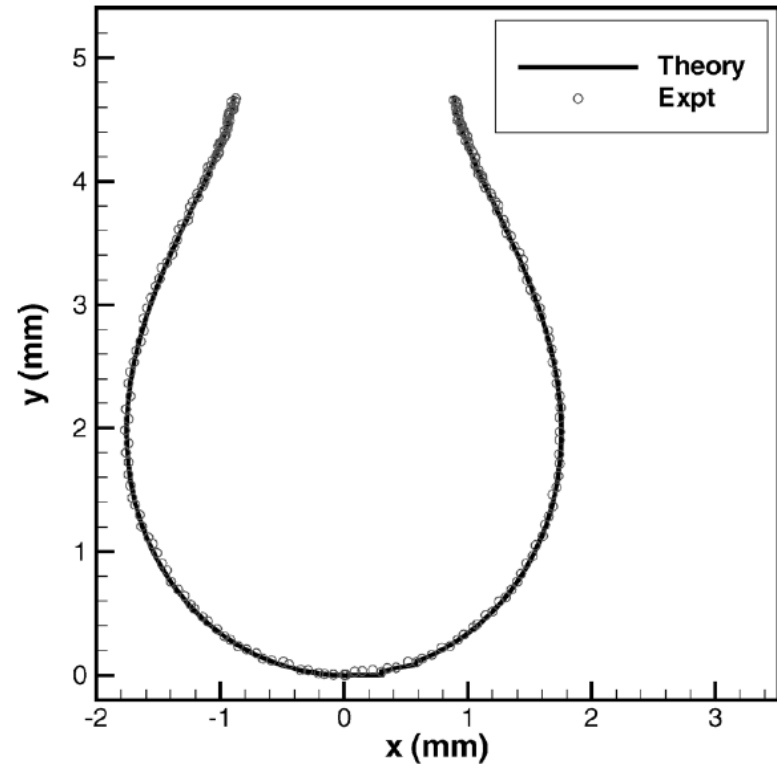
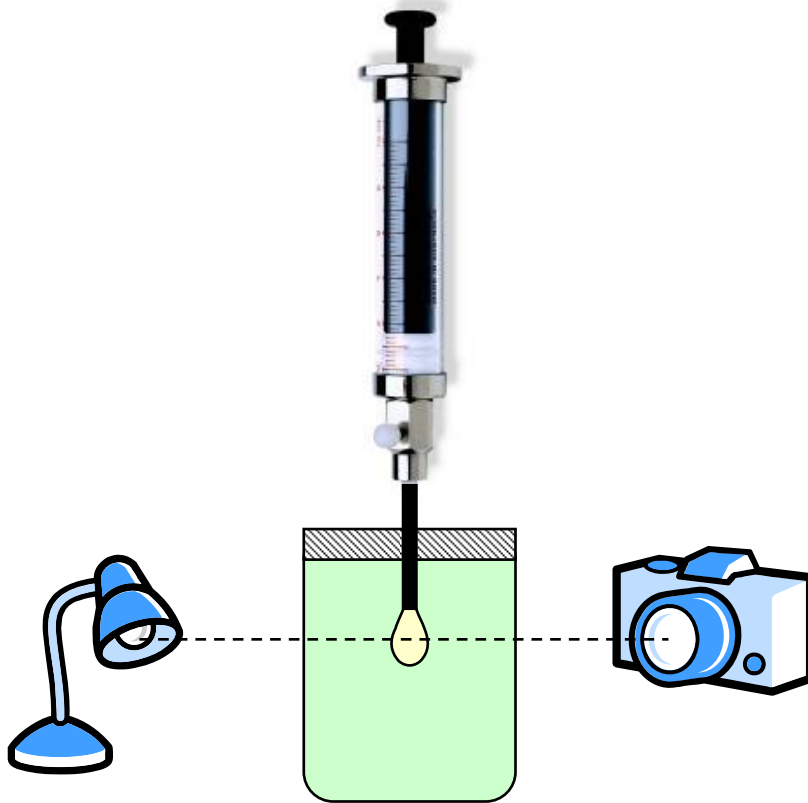
- Pendant Drop
- Wilhelmy Plate
- Maximum Bubble Pressure

Pendant Drop Method

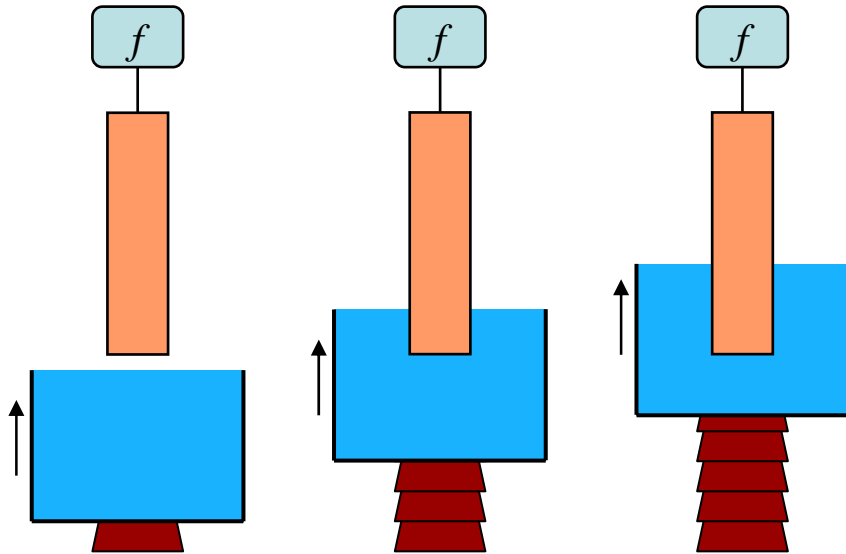
A digital image of the drop silhouette is obtained and fitted with Laplace equation.

The density is an input parameter.

The surface tension is a fitting parameter.



Wilhelmy Plate Method ($\theta = 90^\circ$)

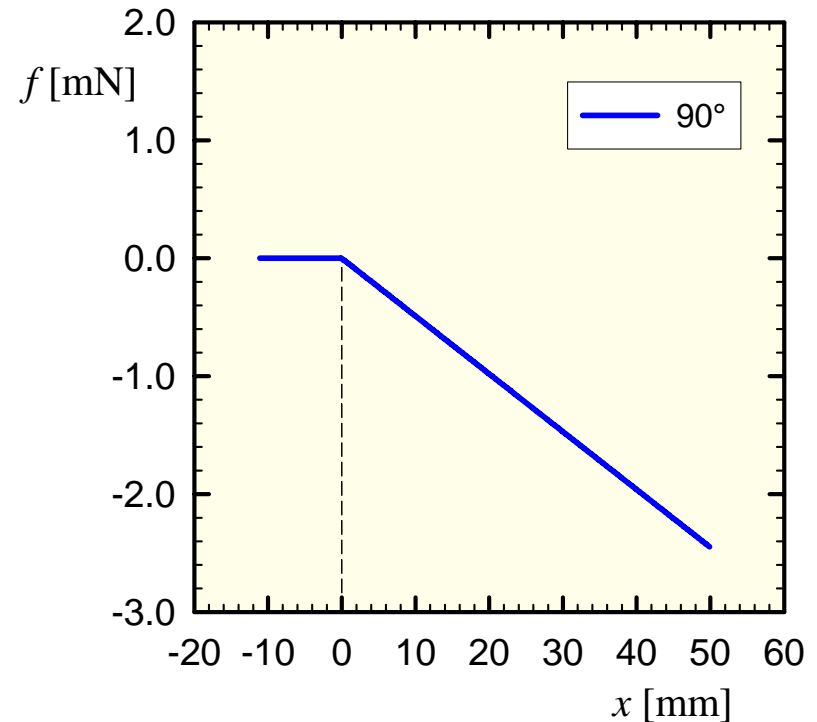


The total force is:

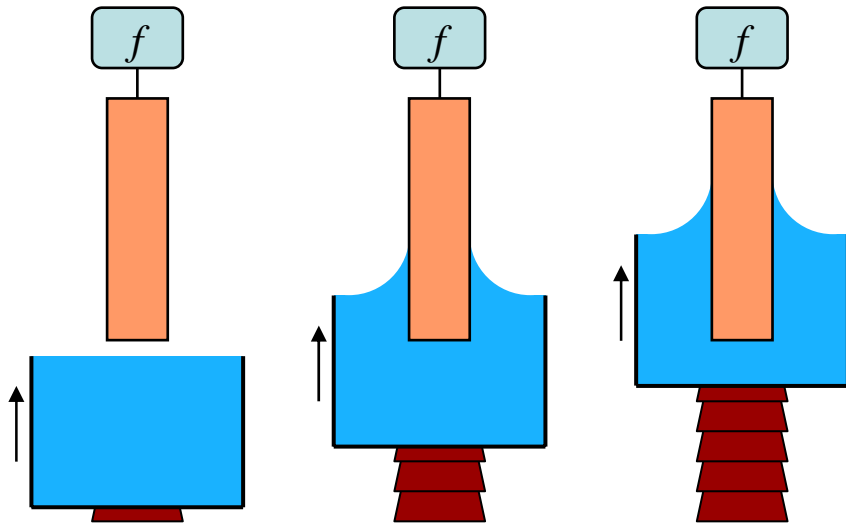
$$\begin{aligned} f_{\text{total}} &= m_0 g - f_{\text{buoyancy}} = \\ &= m_0 g - \rho_L A g x \end{aligned}$$

The recorded force is:

$$f = f_{\text{total}} - m_0 g = -\rho_L A g x$$



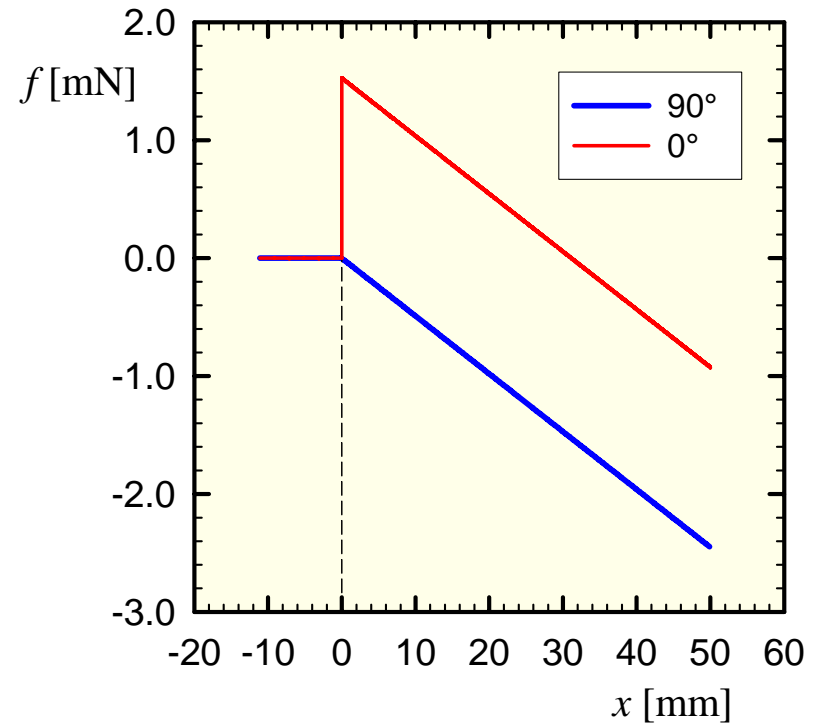
Wilhelmy Plate Method



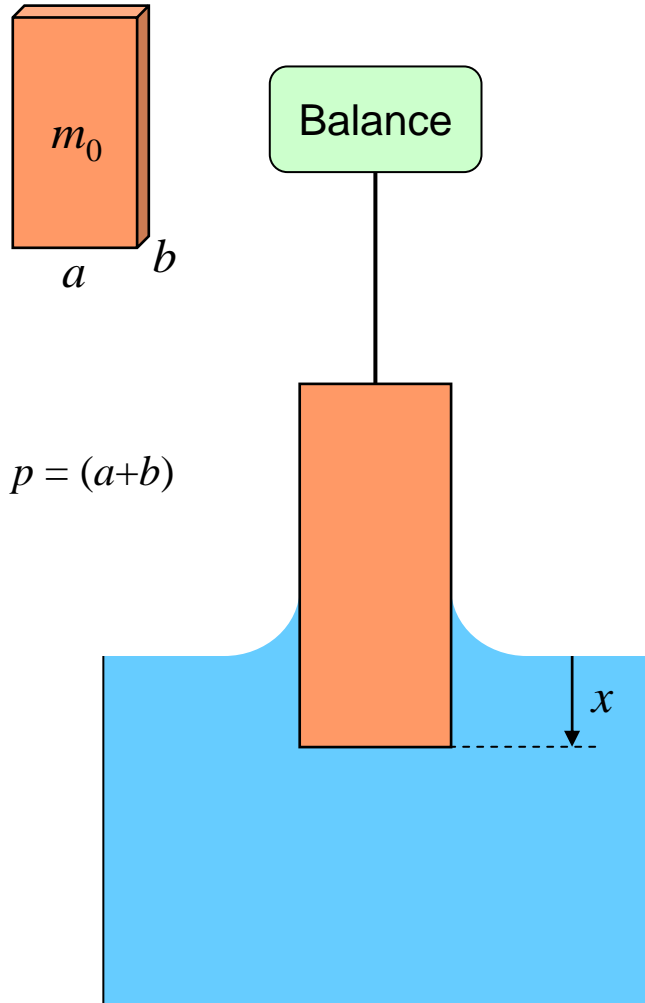
If the contact angle is not 90° there is also a capillary force.

The recorded force is:

$$f = f_{\text{buoyancy}} + f_{\text{capillary}} = -\rho_L A g x + p \gamma \cos \theta$$



Wilhelmy Plate: Calculations



The recorded force is:

$$f = -\rho_L A g x + p \gamma \cos \theta$$

The extrapolated force is:

$$f(x=0) = p \gamma \cos \theta$$

$\theta = 0$

known γ

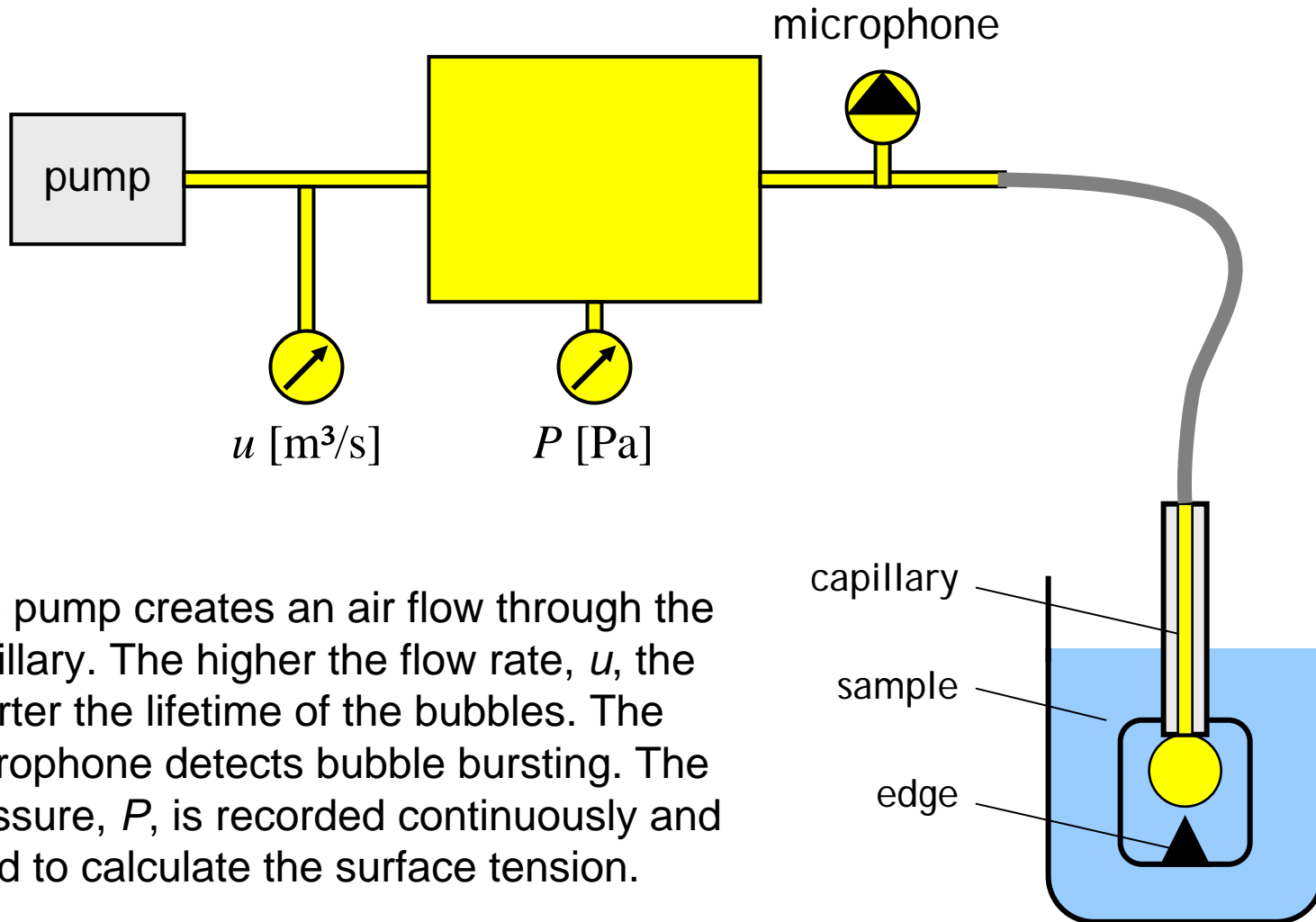
$$\gamma = \frac{f(x=0)}{p}$$

Surface Tension

$$\cos \theta = \frac{f(x=0)}{p \gamma}$$

Contact Angle

Maximum Bubble Pressure Method



The pump creates an air flow through the capillary. The higher the flow rate, u , the shorter the lifetime of the bubbles. The microphone detects bubble bursting. The pressure, P , is recorded continuously and used to calculate the surface tension.

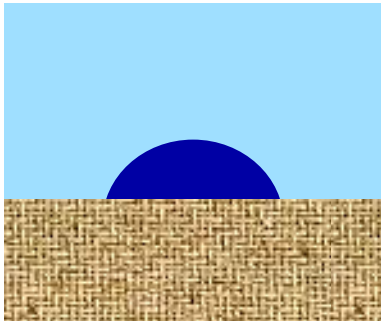
3. Contact Angle on Flat Surfaces

- Sessile Drop Method
- Wilhelmy Plate Method
- Vibrated Contact Angles

Sessile Drop Technique

Protocol:

- Deposit a droplet of liquid on the surface;
- Take an image of the droplet shape;
- Determine the contact angle.

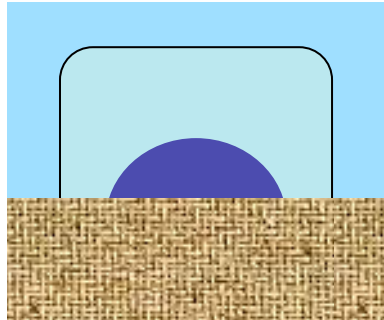


Problems:

- Which angle was measured?
- Was evaporation significant?
- What is the correct droplet size?
- Is the value of θ representative?
- Has the liquid affected the surface?

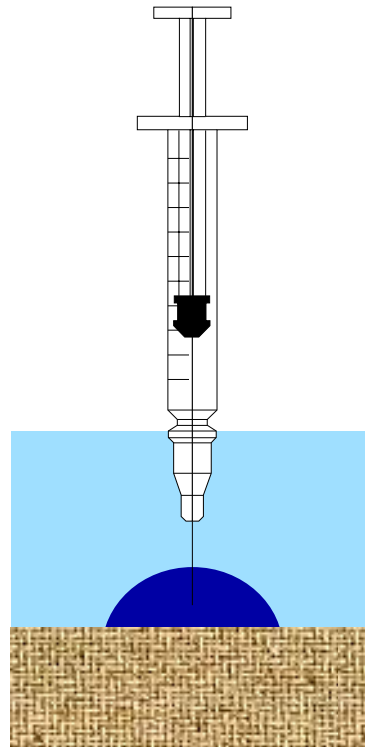
Sessile Drop Technique

Eliminate/reduce evaporation

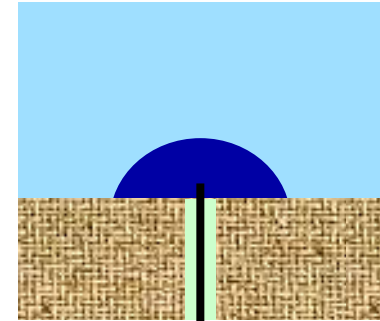


inconvenient

Advance/recede the contact line



most popular



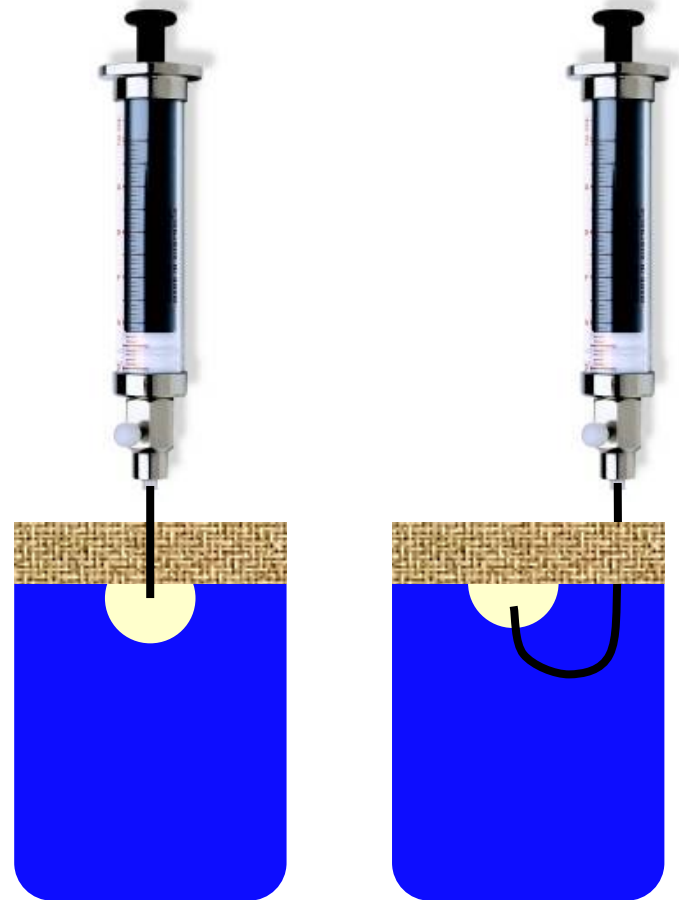
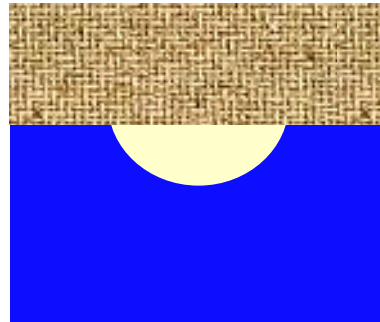
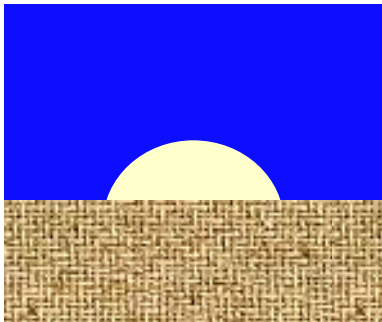
stable/inconvenient

Captive Bubble

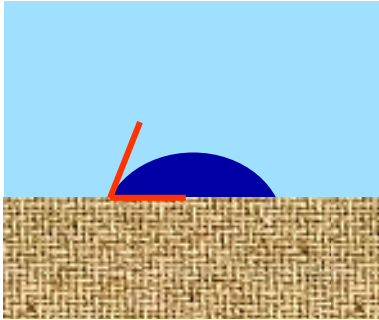
The captive bubble is an “inverted” sessile drop technique.

More complicated but useful when:

- evaporation is important;
- low concentration (e.g. surfactant);
- pH must be maintained.

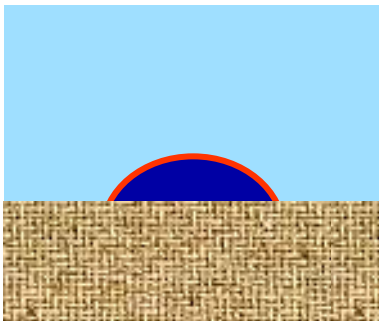


Drop Shape Analysis



Draw the tangent numerically (fit the droplet profile with an empirical function):

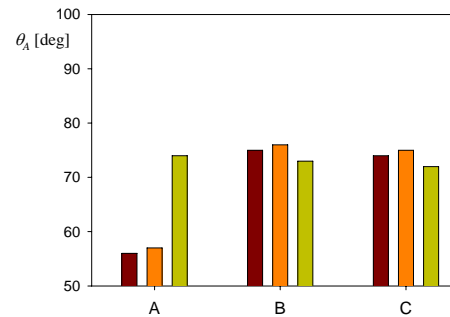
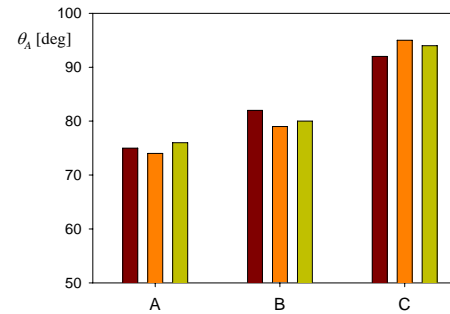
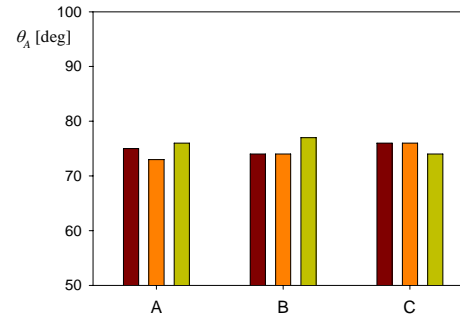
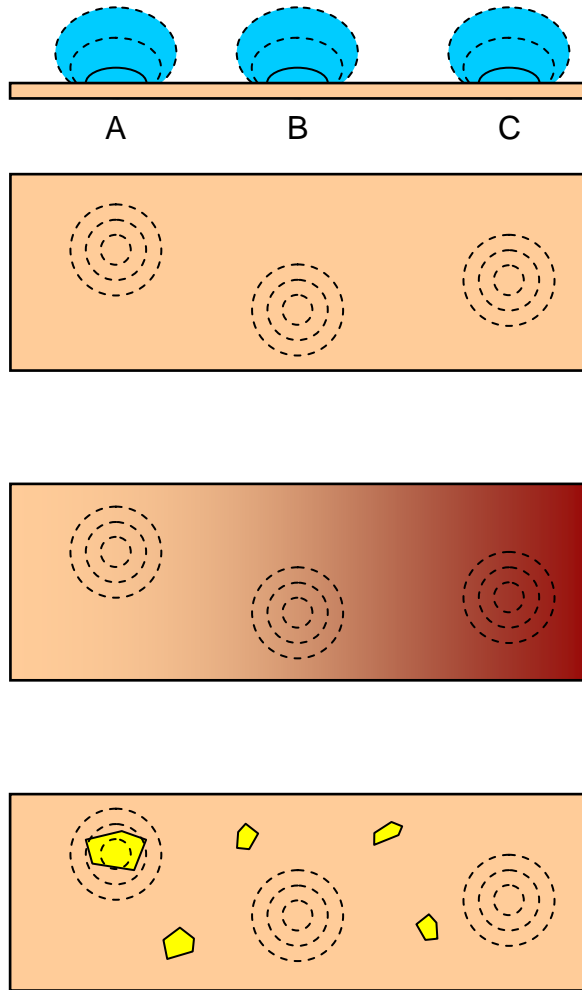
- empirical 🙌;
- tolerates shape asymmetry 👍.



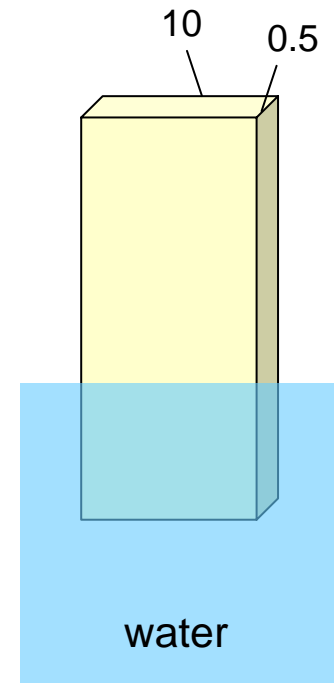
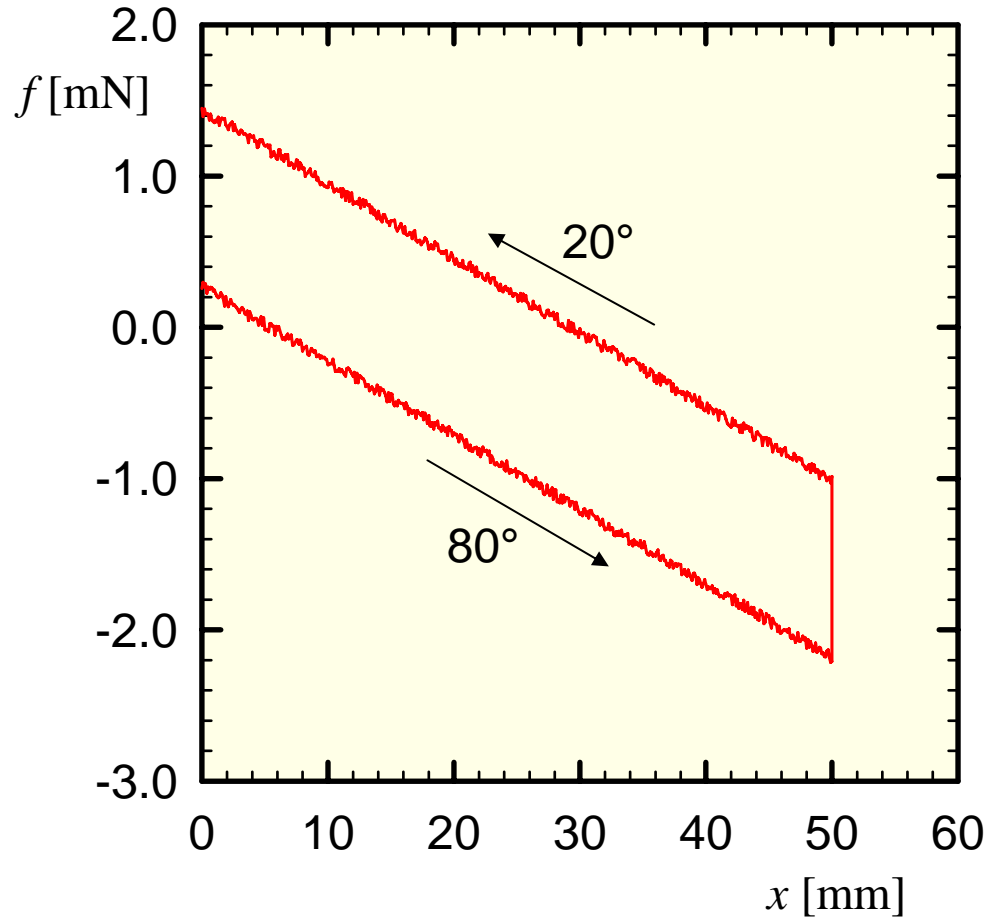
Fit the drop silhouette with Laplace equation:

- conceptually superior 🙌;
- suitable for axisymmetric shapes only 🙌.

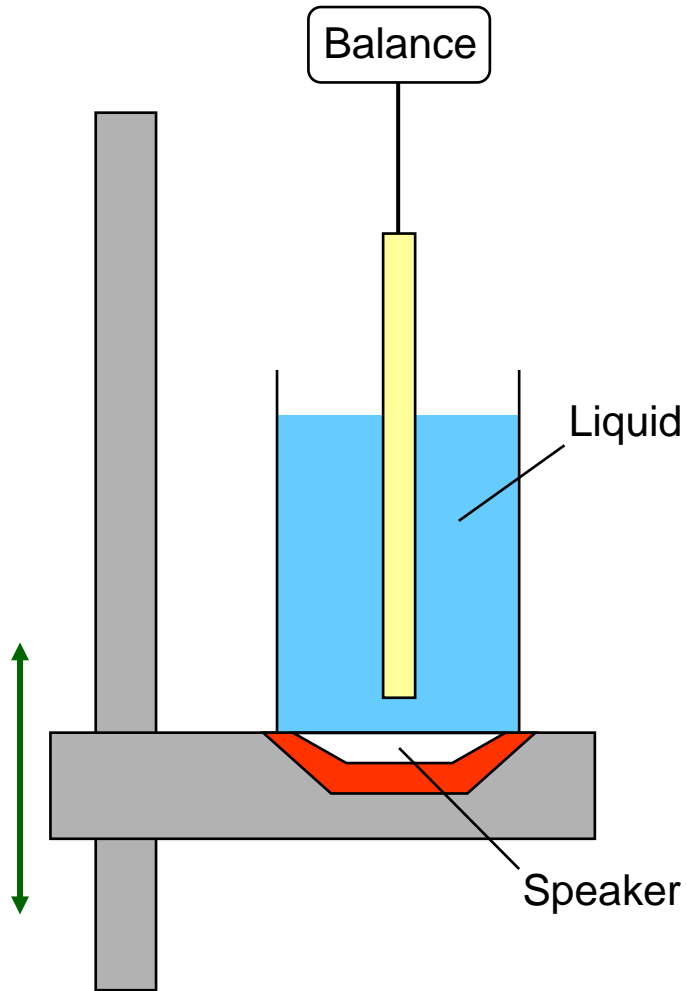
Local vs. Global Measurement



Wilhelmy Plate Method



Wilhelmy Plate with Vibrations

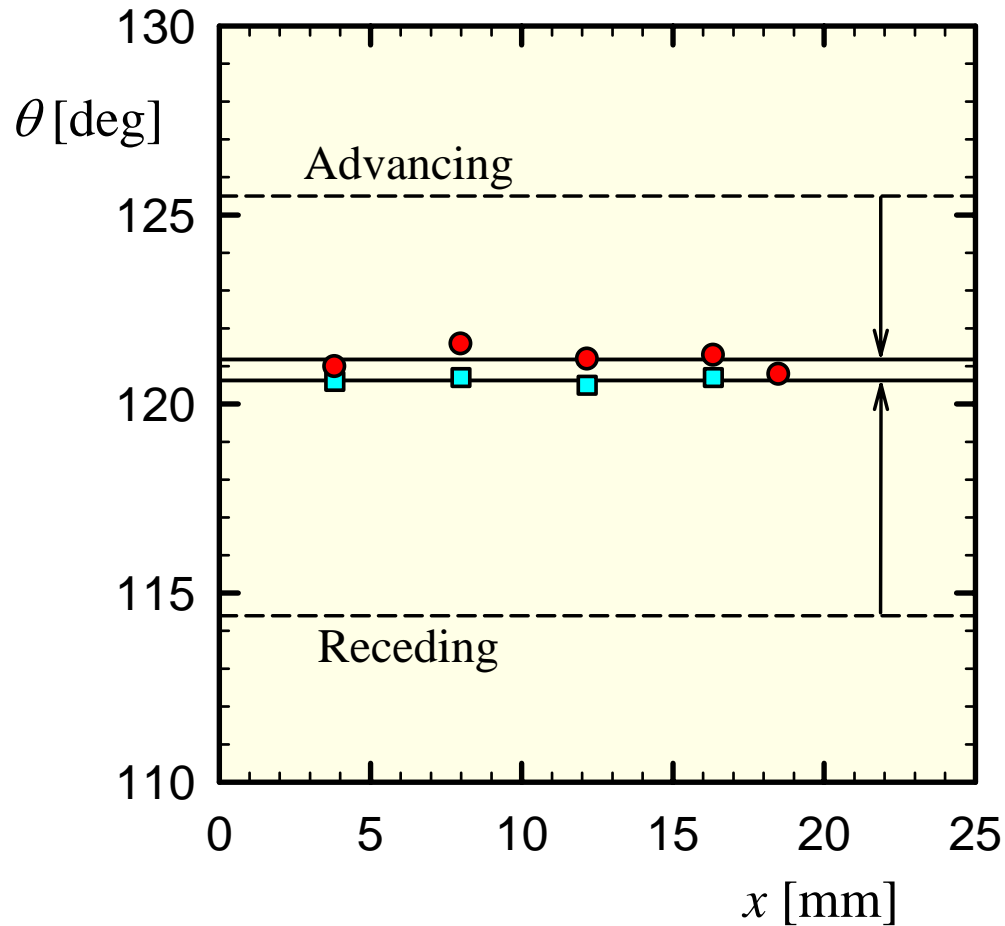


Measurement Protocol:

- drive stage (advance or recede);
- stop movement;
- apply acoustic vibrations;
- allow relaxation to occur;
- measure force.

$$\cos \theta_{A,R,V} = \frac{f_{A,R,V}}{p\gamma}$$

Vibrated Contact Angles



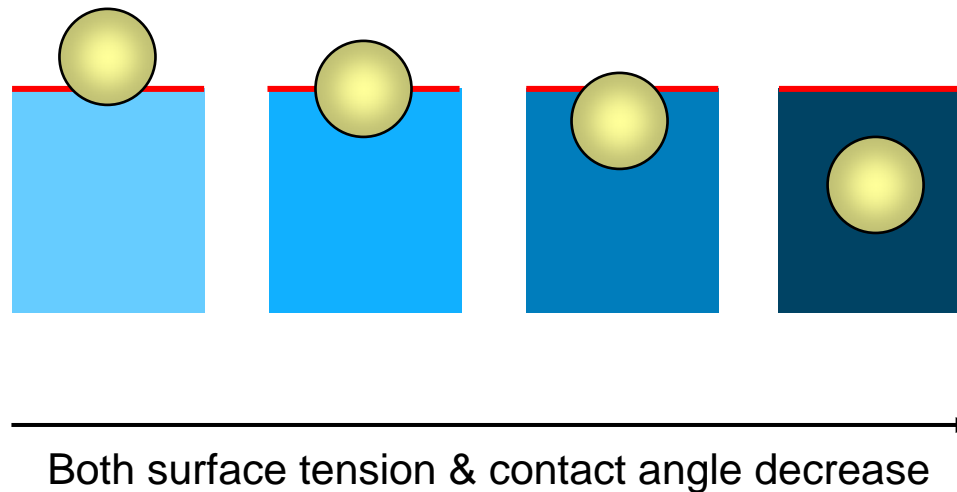
Fabretto et al (2003)

4. Contact Angle on Powders

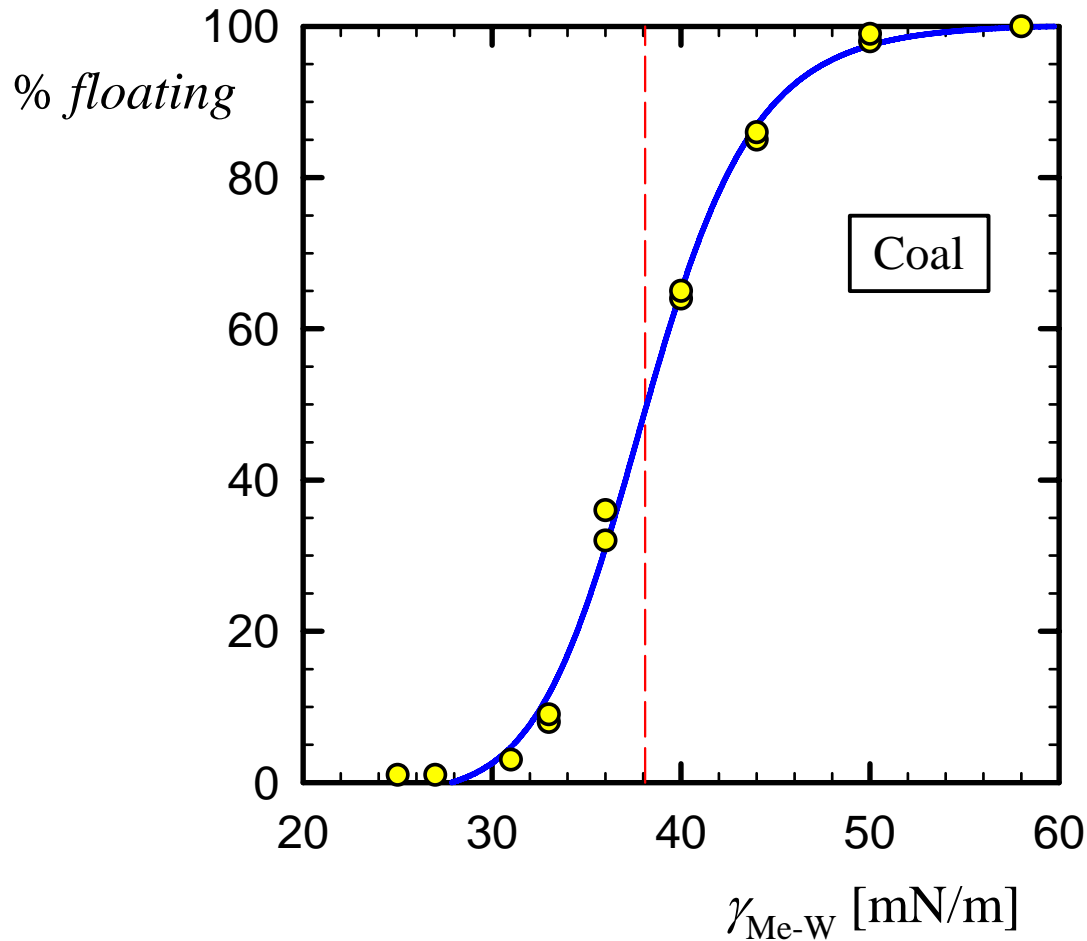
- Skin Flotation
- Static Capillary Pressure
- Washburn Technique

Skin Flotation Technique

Liquid mixtures (e.g. water/methanol) with decreasing surface tension, γ , are consecutively used to determine the critical surface tension, γ_C , of the particle surface.



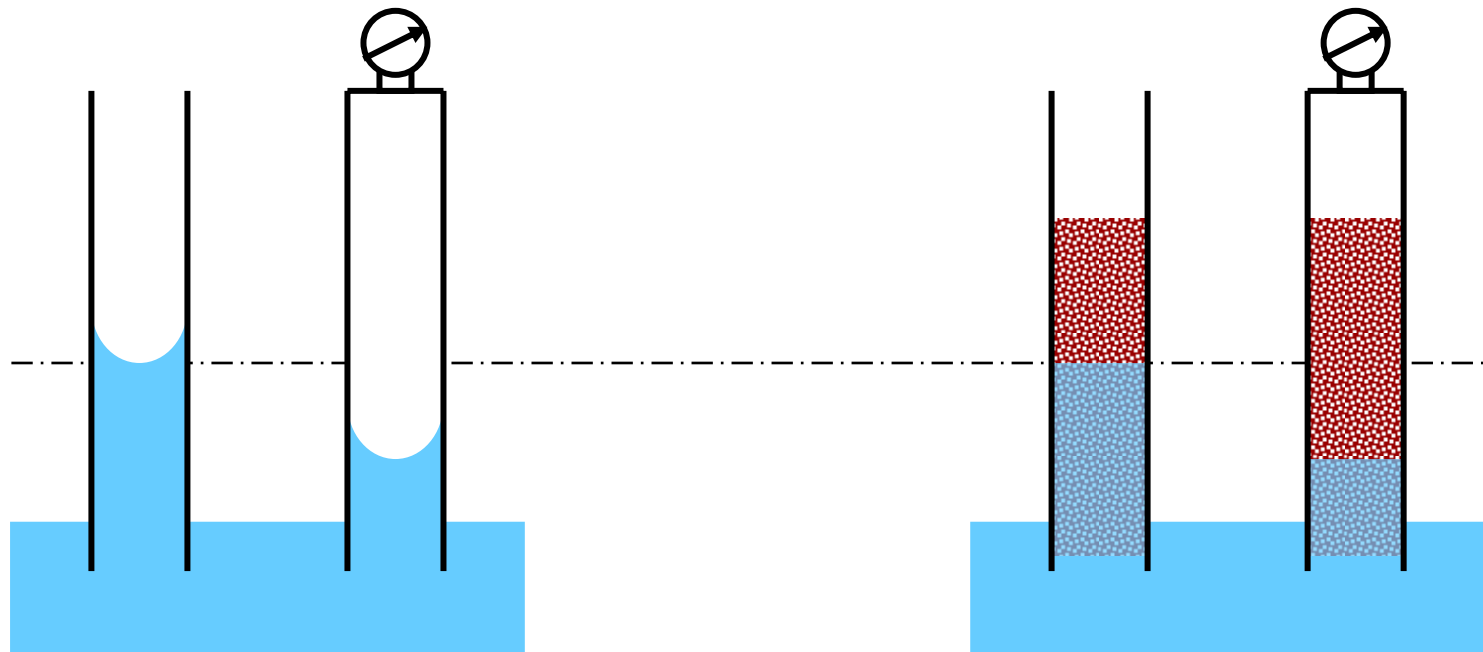
Skin (Film) Flotation – Example



Williams & Fuerstenau (1987)

Static Capillary Pressure Technique (SCPT)

Measure the pressure required to stop capillary penetration:



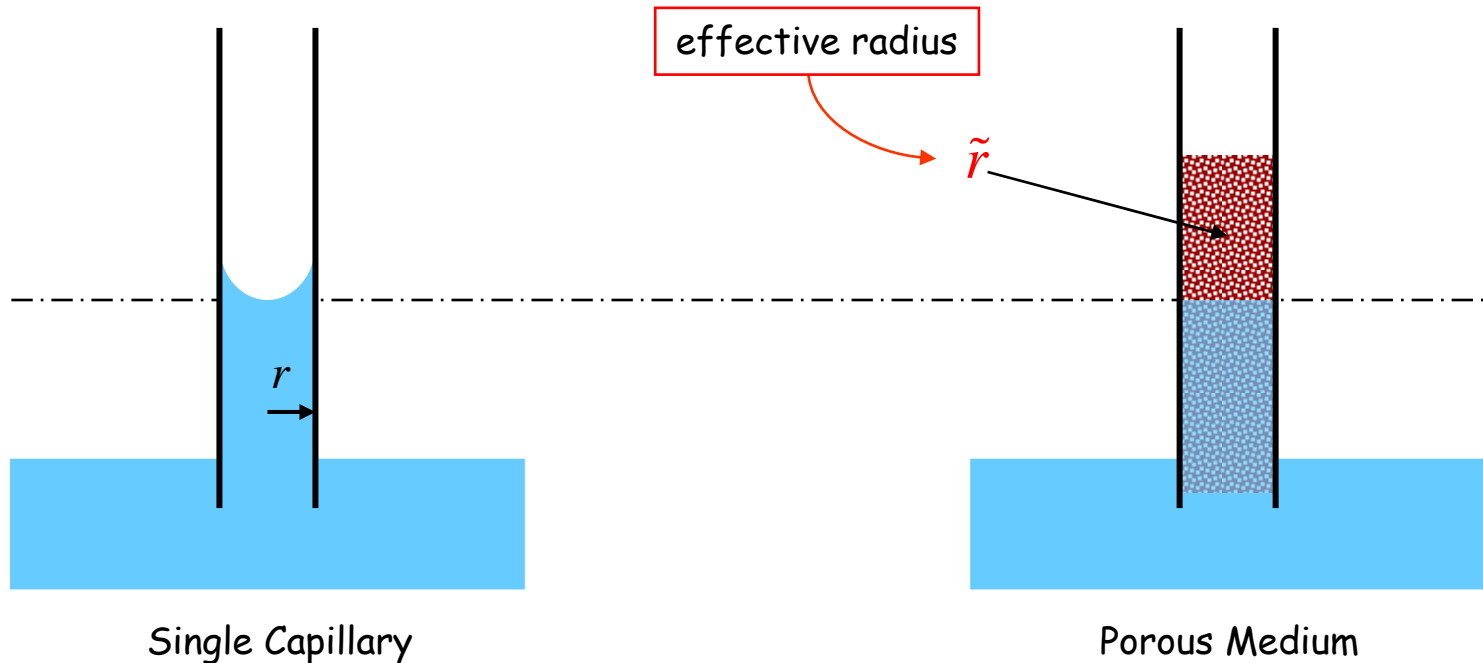
Bartell & Osterhof (1927)

Effective Capillary Radius

The capillary pressure, ΔP , is given by Laplace equation:

$$\Delta P = \frac{2\gamma}{r} \cos \theta$$

$$\Delta P = \frac{2\gamma}{\tilde{r}} \cos \theta$$



SCPT – Advancing Liquid

$$\Delta P = \frac{2\gamma}{\tilde{r}} \cos \theta$$

Test liquid ($\theta_1 > 0$):

$$\Delta P_1 = \frac{2\gamma_1}{\tilde{r}} \cos \theta_1$$

Calibrating liquid ($\theta_2 = 0$):

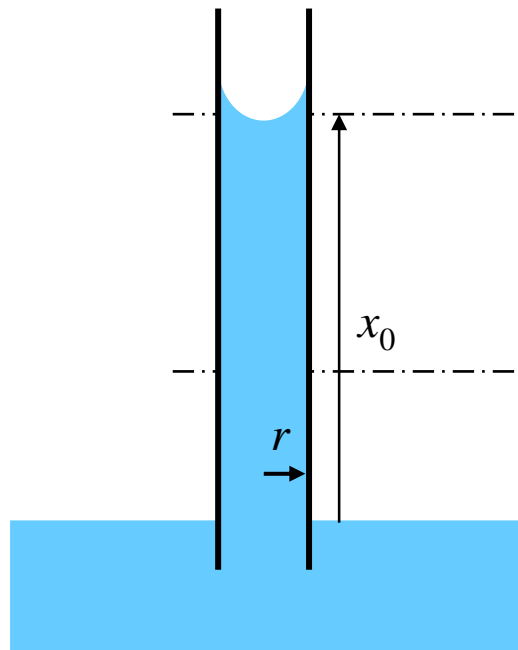
$$\Delta P_2 = \frac{2\gamma_2}{\tilde{r}} \cos 0$$

$$\cos \theta_1 = \frac{\gamma_2}{\gamma_1} \frac{\Delta P_1}{\Delta P_2}$$

Driving Force for Capillary Penetration

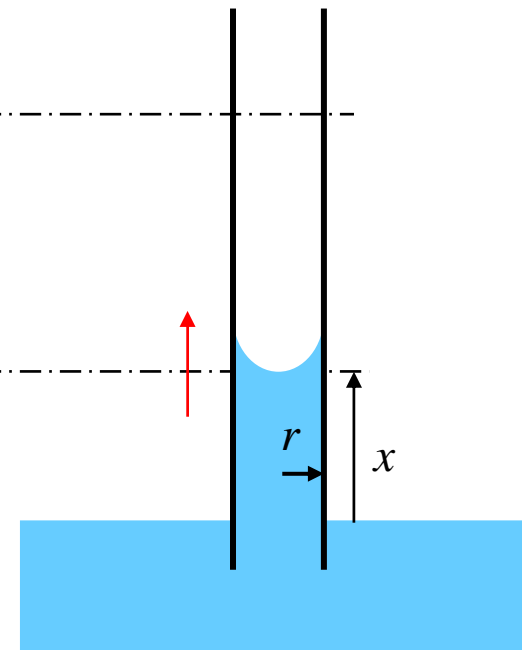
Static Meniscus:

$$f = \frac{2\gamma}{r} \cos \theta - \Delta\rho g x_0 = 0$$



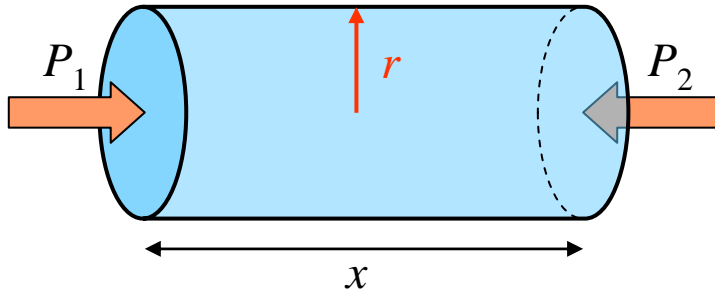
Dynamic Meniscus:

$$f = \frac{2\gamma}{r} \cos \theta - \Delta\rho g x > 0$$



Poiseuille Flow

Circular Pipe

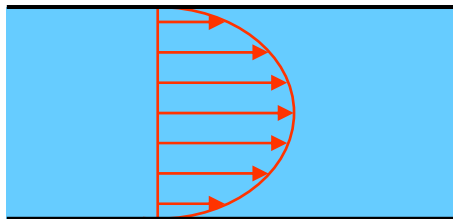


Poiseuille Equation

$$Q = \frac{\pi r^4}{8\mu x} \Delta P$$

$$Q = \frac{V}{t} = \frac{Ax}{t} = A\bar{u} = \pi r^2 \bar{u}$$

Velocity Profile



$$\bar{u} = \frac{r^2}{8\mu x} \Delta P$$

Washburn Equation

Constant capillary pressure

$$\frac{dx}{dt} \equiv \bar{u} = \frac{r^2}{8\mu x} \Delta P \quad \text{Poiseuille Equation}$$

Gravity is unimportant

$$\frac{dx}{dt} = \frac{r^2}{8\mu x} \left(\frac{2\gamma \cos \theta}{r} - \Delta\rho g x \right)$$

Integration

$$\frac{dx}{dt} = \frac{r\gamma \cos \theta}{4\mu x}$$

$$x^2 = \frac{r\gamma \cos \theta}{2\mu} t \quad \text{Washburn Equation}$$

Washburn Method

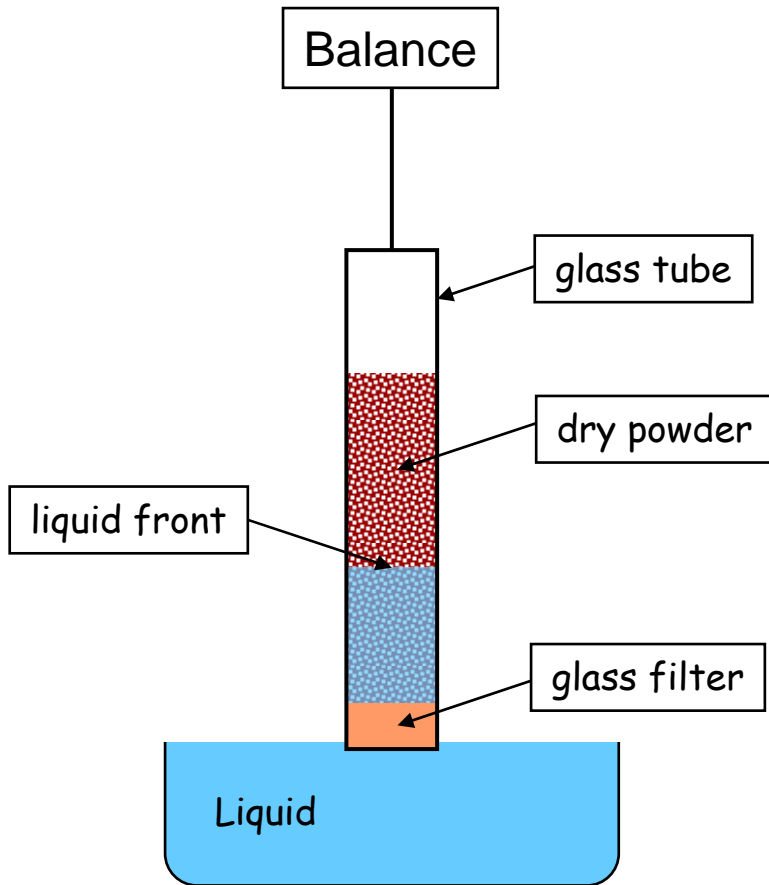
The contact angle is related to the geometry of the system ($2/\tilde{r}$), the known liquid properties (viscosity μ and surface tension γ) and a measurable quantity (x^2/t):

$$\cos \theta = \frac{2 \mu x^2}{\tilde{r} \gamma t}$$

The geometry factor can be calibrated by using a liquid with $\theta_2 = 0$. Then :

$$\cos \theta_1 = \frac{\mu_1 \gamma_2 \left(x^2/t\right)_1}{\mu_2 \gamma_1 \left(x^2/t\right)_2}$$

Washburn Method – Gravimetric Version



Assumptions:

- gravity is ignored;
- flow is laminar;
- pore geometry is constant.

$$\cos \theta = \frac{2}{\underbrace{\varepsilon^2 A^2 \tilde{r}}_{\text{geometry}}} \frac{\mu}{\underbrace{\rho^2 \gamma}_{\text{liquid}}} \frac{m^2}{\underbrace{t}_{\text{experiment}}}$$

Conclusions

- The measuring technique must be regarded as part of the research approach;
- Sensitivity is (potentially) enormous but accuracy is typically low;
- Hysteresis must always be taken into account;
 - Surface may be rough and/or heterogeneous;
 - Surface may not be uniform;
 - Surface may be affected by the contact with the liquid;
 - Surface may be contaminated;
 - Contact line speed may be important (especially with viscous liquids or at high speed);
- Use complementary techniques whenever possible;
- Very few methods work for single particles; measurements on packed beds are often preferable.